Gums, Adhesives & Sealants Technology (with Formulae & their Applications) 2nd Edition
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Naturally occurring polysaccharides from plant exudates have been in use from many decades in immense quantities. Natural gums are natural polymers, which mainly consists of carbohydrates sometimes with small amounts of proteins and minerals. Gum and its derivatives are widely used in various industries as per its needs. The appearance and properties of natural gums determine their commercial value and end use. Due to their extraordinary, unrivalled technological & functional properties gum is used in many industries. Gums not only modify viscosity and consistency, they also often attenuate odour, taste and flavour intensity. Adhesive or sealant is a mixture in a liquid or semi-liquid state that is capable of holding materials together by surface attachment. Adhesives and sealants are used as a raw material for the manufacturing industry or for the service of different processing industries. Adhesives and sealants virtually touch every part of our lives. The adhesives and sealants are two chemically similar but functionally different groups of formulated products. There is no end in sight to the new materials, new formulation, and new uses to which adhesives and sealants will be put in the future.

Some of the fundamentals of the book are advantages of adhesive bonding, hybrids and coupling agents, adhesive films, designing polymers for adhesives, fundamentals of adhesion, designing polymers for adhesives, thermodynamics of adhesion, casein and mixed protein adhesives, lime-free casein adhesives, foil to paper laminating adhesives, casein and protein blend glues as wood adhesives, chemistry of protein blend glues, natural rubber adhesives, vulcanizing latex adhesives, solution adhesives from natural rubber, halogenated butyl rubber, butyl rubber and poly isobutylene lattices, polysulfide sealants and adhesives etc.

This book covers a wide range of polymeric adhesives and sealants, gums along with their essential formularies, distinguished by applications and based on technology. The main areas covered in details are the basic fundamentals, properties, uses and applications, formulations and chemistry, methods of manufacturing and lastly testing methods. This book will be very resourceful to its readers who are just beginners in this field and also to upcoming entrepreneurs, engineers, existing industries, technologist, technical institution etc.

Content:
Specifications Applying to Casein Glue
Viscosity and Pot Life
Certification of Fire Doors
Properties of Casein Glues
Durability of Casein Glues

I STARCH BASED ADHESIVES
MODIFICATION OF STARCHES
Fluidity Starches
Oxidized Starches
Dextrinization
Hydroxyethylation
Cationic Starches
Amphoteric Starches
Miscellaneous Derivatives

EFFECT OF ADDITIVES
Sodium Hydroxide
Borax
Urea
Glycerol
Soluble Soaps
Urea-Formaldehyde Resin
Miscellaneous Additives

STARCH ADHESIVES
Jelly Gums
Other Liquid Formulations
Pastes
Borated Dextrins
White Dextrins
Canary Dextrins
British Gums
Waxy Starch Dextrins
Dextrin/Silicate Blends
Pregelatinized Starches

APPLICATION AREAS
Papermaking
Paper Coating
Corrugating
Bag Adhesives
Laminating Adhesives
Tube Winding
Corrugated Boxes
Gummed Tapes
Label and Envelope Adhesives
Paper Box
Textiles
Wall Covering Adhesives
Miscellaneous Uses
GOVERNMENTAL REGULATIONS: ADDITIVES

I NATURAL RUBBER ADHESIVES
INTRODUCTION
RAW MATERIALS
Natural Rubber Latex
Preservation
Dry Natural Rubber
Natural Rubber Grafted with Methyl Methacrylate (Heveaplus MG)
Depolymerized Rubber
Synthetic Polyisoprene
FORMULATION OF LATEX ADHESIVES
Quick-Grab Adhesive
Self-Adhesive Envelopes
Latex Pressure-Sensitive Adhesives
Tile Adhesives
Reseal Adhesives
Anchor Coat for Tufted Carpets
Other Nonvulcanizing Latex Adhesives
Vulcanizing Latex Adhesives
SOLUTION ADHESIVES FROM NATURAL RUBBER
Nonvulcanizing Adhesives
Vulcanizing Adhesives
Mastics, Asphalts, and Sealants
GRAFTED COPOLYMER HEVEAPLUS MG
FORMULATIONS

I BUTYL RUBBER AND POLYISOBUTYLENE
INTRODUCTION
BASIC PROPERTIES
Butyl Rubber
Polyisobutylene
Halogenated Butyl Rubber
Butyl Rubber and Polyisobutylene Latices
Modified Butyls
FORMULATING AND PROCESSING
Choice of Polymer
Pigments and Fillers
Tackifiers, Plasticizers, and Other Polymeric Additives
Curing Systems
Solvents and Solution Processing
Mixing and Processing Techniques and Guidelines
APPLICATIONS AND FORMULATIONS
Adhesives and Mastics
Sealants
Mixing
I NITRILE RUBBER ADHESIVES
INTRODUCTION
PREPARATION OF NITRILE RUBBER
EMULSION TECHNOLOGY
Functionally Terminated Telechelic Liquid Polymers
COMPOUNDING NITRILE RUBBER CEMENTS
Polymer Selection and Solubilization
Types of Compounding Ingredients
APPLICATION
Nitrile Rubber Systems
Nitrile Rubber/Phenolic Adhesives
Nitrile Rubber/Epoxy Adhesives

I STYRENE-BUTADIENE RUBBER ADHESIVES
INTRODUCTION
Perspective
History of SBR
Manufacture of SBR
Basic Chemistry of SBR
SBR LATEXES IN ADHESIVES
General
Classification
Benefits of SBR Latexes
Compounding Ingredients
Major Applications
SBR (SOLID) IN ADHESIVES
General
Classification
Compounding Ingredients
Major Applications

I NEOPRENE (POLYCHLOROPRENE) -
BASED SOLVENT AND LATEX ADHESIVES
HISTORY
THE EFFECT OF POLYMER STRUCTURE
NEOPRENE SOLVENT-BASED ADHESIVE CEMENTS
Types of Neoprene
Antioxidants
Metal Oxides
Resins
Fillers
Curing Agents
Solvents
Adhesive Processing
End Uses
Application Methods
NEOPRENE LATEX-BASED ADHESIVES
Anionic Types
Nonionic Type
Compounding
Typical Formulations

I POLYSULFIDE SEALANTS AND ADHESIVES
POLYSULFIDE SEALANTS
Chemistry
Compounding
Curing Agents
Fillers
Plasticizers
Adhesion
Primers
Specifications
ADHESIVES FROM POLYSULFIDE LIQUID
POLYMER-EPOXY RESIN REACTIONS
Chemistry
Physical Properties
Applications
OTHER MERCAPTAN-TERMINATED POLYMERS
Polyethers
Polyesters
Urethanes
Olefin

I PHENOLIC RESIN ADHESIVES
CHEMISTRY
Formaldehyde
Novolaks
Strong Acid Reactions
Weak Acid Reactions
Resoles
Dispersion Resoles
Resin Cure
MANUFACTURE
ABRASIVES
Bonded Abrasives
Coated Abrasives
COATINGS
FOUNDRY
FRICITION COMPOSITES
MOLDING COMPOUNDS
PHOTORESISTS AND CARBONLESS PAPER
LAMINATING
WOOD BONDING
INSULATION AND FOAM
GENERAL ADHESIVES
ENVIRONMENTAL AND TOXICOLOGICAL CONSIDERATIONS
I AMINO RESIN ADHESIVES
HISTORY
RAW MATERIALS
CHEMISTRY
END USES
Particleboard
Plywood
High-Pressure Decorative Laminates
Miscellaneous Applications
TOXICITY

I EPOXY RESIN ADHESIVES
INTRODUCTION
EPOXY RESINS USED IN ADHESIVES
Bisphenol A Based Epoxy Resins
Epoxy Novolac Resins
High performance Epoxy Resins
Flexible Epoxy Resins
CURING AGENTS USED IN ADHESIVES
Polysulfide
Amines
Aliphatic Amines
Cycloaliphatic Amines
Aromatic Amines
Polyamides
Amidoamines
Dicyandiamide
Catalytic Curing Agents
Anhydrides
SUMMARY
DILUENTS
FILLERS
ELASTOMERIC MODIFIERS
TYPICAL ADHESIVE FORMULATIONS
COMPLEMENTARY TECHNOLOGIES
Robotics
Induction Curing
APPLICATIONS AND SUMMARY

I POLYURETHANE - AND ISOCYANATE -
BASED ADHESIVES
REASONS FOR THE EFFECTIVENESS
OF POLYURETHANE AND ISOCYANATE-BASED
ADHESIVES 321-323
TYPES AND USE OF POLYURETHANE
AND ISOCYANATE BASED ADHESIVE SYSTEMS
Method A (isocyanate primer)
METHOD B (CONVENTIONAL PLASTIC OR
RUBBER VEHICLE + ISOCYANATE)
The Relative Effectiveness of "Vulcabond" T and Tx in
Rayon Cord-to-Natural Rubber Adhesion
Method C (in situ polyurethane polymerization)
Method D (polyurethane elastomer without or
with added polyisocyanate)
Method E (blocked di- or polyisocyanate)
Method F (aqueous dispersion)
Method G (film and tape)
Method H (powder)
POLYURETHANE STABILIZATION
HANDLING ISOCYANATE BASED ADHESIVES
IDENTIFICATION OF ADHESIVE COMPONENTS

I. POLYOLEFIN AND ETHYLENE COPOLYMER-BASED
HOT MELT ADHESIVES
ADHESIVE FORMULATION
Polymers
Tackifiers
Waxes
HOT MELT ADHESIVES APPLICATIONS
Cases, Cartons, and Trays
Bookbinding
Nonwovens
Furniture
Labels
Polyester Beverage Bottles
Carpet Seaming Tape
Paper Laminates
HOT MELT APPLICATION EQUIPMENT
FUTURE OF HOT MELT ADHESIVE

I. POLYVINYL ACETAL ADHESIVES
CHEMISTRY
HEALTH, TOXICOLOGY, AND SAFETY
PHYSICAL PROPERTIES
Solubility
Compatibility
Viscosity
Mechanical Properties
Thermal Properties
USES AS AN ADHESIVES
Hot Melts
Thermosetting Adhesives
Adhesion to Metal
Adhesion to Natural Surfaces
Green Strength Binder
Composites
Adhesion to Glass
Other Uses

I ACRYLIC ADHESIVES
TECHNOLOGY
Chemistry
Glass Transition Temperature (Tg)
Crosslinked Thermosets
BONDING PROCESSES
Pressure Sensitive
Contact Bonding
Heat and Pressure Bonding
Vacuum Bonding
Wet Laminating Adhesives
Filled Adhesives
ENGINEERING ADHESIVES
Chemistry and Technology
Radiation Curing
Application Processes
Current Uses

I PRESSURE-SENSITIVE ADHESIVES FOR TAPES AND LABELS
CONSTRUCTIONS
Manufacture
Backings
Release Coatings and Liners
ADHESIVE SYSTEMS
Tackifiers
Rubber-Based Adhesives
Acrylics
Silicones
Miscellaneous Polymers
Surface Energetics
Tack
Peel Adhesion
Cohesive Strength
TEST METHODS
Tack Testing
Peel Adhesion Testing
Shear Resistance Testing
Miscellaneous Tests
Tapes
Labels and Decals
Other Products
Adhesive Mass Thicknesses

I FACTORS INFLUENCING GUM COSTS AND APPLICATIONS
DEFINITION OF GUM
FACTORS AFFECTING RAW GUM COSTS
1. Exudate Gums
2. Seaweed Gums
3. Seed Gums
4. Starch and Cellulose Derivatives

INDUSTRIALLY VALUABLE PROPERTIES OF GUMS
1. Linear Neutral Polysaccharides
2. Brached Neutral Polysaccharides
3. Polysaccharides with Carboxyl Groups
4. Polysaccharides with Strong Acid Groups
5. Polysaccharides with basic Groups

MODIFIED GUMS
1. Introduction of Neutral Groups
2. Introduction of Acideic Groups
3. Introduction of Groups
4. Graft Polymers
5. Other Chemical Modifications of Natural Polysaccharides

I AGAR
INTRODUCTION
SOURCE
1. Raw Material
2. Processing
3. Finished Product

HISTORY
1. Discovery
2. Manufacture
3. Use
4. Present Applications
5. Derivatives

IV. STRUCTURE
PROPERTIES
1. Solid
2. Sols
3. Gels

I ALGIN
INTRODUCTION
PRODUCTION
1. Source
2. Seasonal Variations
3. Harvesting
4. Processing
5. Grades
6. Industrial Importance
7. Potential Amount
APPLICATIONS
1. Introduction
2. Mode of Action
3. Foods
DAIRY PRODUCTS
BAKERY PRODUCTS
OTHER FOOD PRODUCTS
4. Pharmaceuticals and Cosmetics
5. Industrial Applications
PAPER PRODUCTS
TEXTILE PRODUCTS
RUBBER PRODUCTS
OTHER INDUSTRIAL USES
STRUCTURE
PROPERTIES
1. Dissolution
2. Solution Properties
3. Gels
4. Films
5. Compatibilities
6. Algin in Foods
7. Other Properties

I LAMINARAN
INTRODUCTION
PRODUCTION
1. Source
2. Producing Areas
3. Seasonal Effects
4. Collection
5. Preparation
HISTORY OF DEVELOPMENT
STRUCTURE
PROPERTIES
CHITIN AND ITS DERIVATIVES
INTRODUCTION
PRODUCTION
1. Crustaceans
2. Insects
3. Fungi
4. Preparation
5. Preparation of Deacetylated Chitin, Other Chitin Derivatives, and D-Glucosamine
6. Grades
7. Potential Amount
HISTORY
IV APPLICATION
1. Sizing
2. Adhesives
3. Emulsion Stabilization and Thickening
4. Pharmaceuticals and Cosmetics
5. Extruded Fibers and Films
6. Glycosamine Hydrochloride in Foods and Pharmaceuticals
STRUCTURE
PROPERTIES
1. Solubility and Viscosity
2. Gels
3. Films and Fibers
4. Adhesiveness
5. Compatibilities

GUM ARABIC
INTRODUCTION
PRODUCTION
1. Republic of the Sudan
2. Nigeria
3. Tanganyika
4. Morocco
5. British Somaliland and Abyssinia
6. South Africa
7. India
8. Australia
9. Miscellaneous
10. Types of Gum in the United States
STRUCTURE
ARABIC ACID
1. Preparation
2. Properties
3. Degraded Gum Arabic
4. Derivatives of Arabic Acid

FACTORS WITH AFFECT VISCOSITY
1. Concentration
2. Temperature
3. Electrolytes
4. pH
5. Solvents Others Than Water
6. Aging
7. Mechanical Treatment
8. Ultrasonic Vibrations and Ultravilent Irradiation

OTHER PHYSICAL PROPERTIES
1. Surface Tension
2. Freezing Point

COACERVATION
1. Introduction
2. Gum Arabic-Gelatin Coacervates
3. Preparation of Coacervates
4. General Properties, Physical Appearance, and Composition
5. Effect of Temperature
6. Effect of pH
7. Reactions of Salts
8. Physical Phenomena
9. Uses of Gum Arabic-Gelatin Coacervates
10. Coexisting Coacervates
11. Other Coacervates

CHEMICAL PROPERTIES
1. Chemical Reactivity
2. Solubility
3. Enzymes

SEPARATION AND IDENTIFICATION OF GUM ARABIC
1. Isolation of Gum Arabic from Commercial Products
2. Systematic Analytical Scheme
3. Physical Confirmatory Tests
4. Chemical Confirmatory Tests
5. Direct Tests for Gum Arabic in Some Commercial Products

USES
1. Foods
2. Adhesives
3. Paints
4. Inks
5. Lithography
6. Textiles
7. Miscellaneous

CORN HULL GUM

INTRODUCTION
SOURCE AND PREPARATION
DEVELOPMENT OF USE
STRUCTURE
PROPERTIES

1 GUAR GUM
INTRODUCTION
PRODUCTION
1. Source and Producing Areas
2. Agronomics
3. Purification
4. Grades
USES
1. HISTORY
2. Mining Industry
3. Foods
4. Cosmetics and Pharmaceuticals
5. Paper Industry
6. Explosives
7. Derivatives
STRUCTURE
PROPERTIES
1. Viscosity
2. Gels
3. Films
4. Adhesiveness
5. Miscellaneous

1 GUM KARAYA
INTRODUCTION
Production
1. Source
2. Producing Areas
3. Seasonal Effect
4. Collection
5. Purification
6. Grades
7. Impurities
8. Potential Amount
USES
1. History
2. Commercial Value
STRUCTURE
PROPERTIES
1. Dissolution and Viscosity Measurements
2. Gels
3. Films
4. Adhesiveness
5. Acid Resistance
6. Dispersibility
7. Ropiness
8. Water Retention

PECTIN

INTRODUCTION

PRODUCTION

1. Source
2. Producing Areas
3. Seasonal Effects
4. Collection
5. Purification
6. Grades and Definitions of Grades
7. Impurities
8. Quantities Marketed
9. General Industrial Uses Other Than in Foods

FOOD

Mode of Action

Structure

Properties

1. Pectin Types as Defined by Degree of Methylation
2. Solubility
3. Solutions, Stability and Viscosity
4. Gels
5. Assay Methods, Calculations, and Composition

DIRECTORY SECTION

PRESENT MANUFACTURERS
SUPPLIERS OF RAW MATERIALS
SUPPLIERS OF THE PLANT M/C & EQUIPT.

Sample Chapter:
Starch Based Adhesives

Starch is a natural polymer, available in very large quantities and at relatively low and stable prices. It consists of glucose units chemically bound together so as to form a nonreducing polyhydroxy material. Because of the many hydroxyl groups, starch has a high affinity for polar substances such as water or cellulose. Starch can be reduced to low molecular weight sugars by enzymes called amylases, or by acid hydrolysis.

Some major changes in starch usage in adhesives and related fields are the large increase of cationic and amphoteric starches in paper manufacture and the increased use of high amylose starches in the corrugating industry.

The increased emphasis on recycling makes the use of starch desirable because amylases are able to degrade the starch with essentially no effect on the major part of the adherent matrix. There is a patent on the use of amylase as a release agent in starch glued materials recycling.

Most of the starch used in adhesives in the United States is produced from corn or maize. There are four commercially available starch types that are used in adhesives. These are waxy corn starch, regular corn starch, high amylose type V corn starch, and high amylose type VII corn starch. The major difference between these starches is in the amount of amylose contained in them—approximately 0, 28, 55, and 70%, respectively, by iodine titration. Other starches usable in adhesives include sorghum starch, potato starch, tapioca starch, wheat starch, rice starch, and sago starch. The term sago was originally applied to starch from the stem of Metroxylon-type palm trees, but is often used to describe starches from other palm trees, or even applied to some varieties of tapioca starch. Some characteristics of several commercially available starches are shown in Table 1.

Regular corn starch consists of two major fractions that can be separated by precipitation with butanol under appropriate conditions. The fraction that precipitates out is called amylose. Amylose is essentially linear in form, stains blue with iodine solutions, tends to form a rigid gel from concentrated solutions or to precipitate from dilute solutions, and is about 95% digested by beta amylase (a test for strict linearity or lack of branching) when very carefully isolated. Amylose forms strong, water-resistant films when a solution is evaporated.

The other main fraction is called amylopectin. Amylopectin is highly branched (one branch every 14 to 27 glucose units), stains brown to purplish with iodine solutions, tends to remain in solution at room temperature, and is digested to about 55% by beta amylase. Amylopectin forms weak, water sensitive films.

Regular corn starch (when carefully fractionated) contains an additional intermediate fraction. This fraction is precipitated out of the original corn starch solution by butanol, but is not reprecipitated when water and butanol are added to the amylose fraction dissolved in dimethylsulfoxide. The intermediate fraction is precipitated by iodine from the various solutions, and is 5-7% of the starch in regular cornstarch. The full analysis for regular cornstarch is 25-27% fractionated amylose, 68% amylopectin, and 5-7% intermediate fraction.

Table 1. Commercial starches. Approximate Data and Ranges.

<table>
<thead>
<tr>
<th>Starch</th>
<th>Corn</th>
<th>Wheat</th>
<th>Rice</th>
<th>Tapioca</th>
<th>Potato</th>
<th>Sago</th>
</tr>
</thead>
<tbody>
<tr>
<td>Source</td>
<td>Seed</td>
<td>Seed</td>
<td>Seed</td>
<td>Root</td>
<td>Root</td>
<td>Pith</td>
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When waxy corn starch is carefully fractionated, 2% intermediate fraction is found. The remainder is amylopectin.

High amylose corn starches differ from regular corn starch in that the high amylose corn starches contain more amylose, much more intermediate fraction and much less amylopectin than regular corn starch. For example, a high amylose type VII starch having 70% amylose by iodine titration contained 63% recrystallized butanol complex (standard amylose), 31% intermediate fraction, and only 5% amylopectin. About half of the intermediate fraction is probably low molecular weight amylose (19,000 daltons). This low molecular weight amylose fraction may be responsible for some of the rapid bond formation noted in corrugating adhesives using high amylose starches. The standard amylose determination (70%) is not fully indicative of possible performance. A better analysis would be: normal amylopectin 5%; low molecular weight, less branched amylopectin about 16%; high molecular weight amylose 63%. Total amylose is about 79%, and total amylopectin is about 21%.

Some of the true solution properties of starch appear to be related to the molecular weight of the amylose or amylopectin fractions. For example, the viscosity of jelly gums probably depends on the amylopectin molecular weight. The literature has tended to show increased molecular weights as the methods of isolating the fractions of starch and the methods of determining molecular weight have improved. Most literature indicates that amylose has a molecular weight of about 1,000,000 to 2,000,000 depending on its source, and amylopectin has a molecular weight of about 400,000,000. These are weight average molecular weights determined by light scattering experiments of the fractions with the starch dissolved in dimethylsulfoxide. Commercially used starches probably have much lower molecular weights. For example, the amylopectin of one cultivar of potato starch was consistently determined to be 65,000,000, while the amylopectin from a second cultivar was just as consistently determined to be 440,000,000. A sample of waxy maize starch had a molecular weight of 400,000,000 but after shearing a 15% solution in boiling water the amylopectin molecular weight was determined to be 10,000,000. At high molecular weights the error of measurement becomes quite large. The error at 100,000,000 is Å±10%, and at 500,000,000 the error is Å±20%. The amylopectin from pea starch appears to be the highest recorded, being 1,500,000,000 in one case.

The terms linear and branched were used in the preceding discussion in describing amylose and amylopectin. Fig. 1 illustrates the way in which glucose units are linked together to form starch. Most of the glucose units in starch are linked in a (1-4) alpha-D-linkage. All of the linkages are of this type in amylose (theoretically at least). This is called the linear polymer. Amylopectin contains, in addition to the above linkage, a (1-6)-alpha-D-linkage once in about 25 glucose units. The 1-6 linkage is called a branch point and the linear extension of this branch point is called a branch. Amylopectin is therefore called a branched polymer. Many amylases are able to hydrolyze at both the 1-4 and the 1-6 linkages, but beta amylase
hydrolyzes only at the 1-4 linkage. Further, beta amylase is able to attack only at the nonreducing end of the starch molecule, and hydrolyzes off one two-glucose unit at a time only if that units is linked 1-4 (maltose). Hence, the low digestion by beta amylase (65% in some commercial amylose) suggests an occasional short branch in all except the most highly purified samples.

When starch is suspended in water it tends to increase in volume and to absorb about its weight of water. If the temperature is slowly increased there is a point at which water absorption increase dramatically. The starch granule expands 10-100 times in volume depending on the type of starch. If the amount of starch present is greater than required to absorb all of the water available, then the viscosity of the starch dispersion becomes very heavy. The swollen granules can be partially destroyed by mechanical agitation, and the viscosity will decrease, depending on the type of starch. The temperature at which the starch suddenly swells when heated in water is called the gelatinization temperature.

The unswollen starch granule is in a crystalline stated and as such is anisotropic. Most of the starch granules are birefringent under the polarizing microscope, showing a "polarization cross." These disappear at or near the gelatinization temperature when heated in water, indicating a loss of crystallinity. With further heating the granules tend to swell a little further, collapse to a degree, and fragment if sheared to any extent. Although some starches such as waxy starch, potato starch, tapioca starch, and sago starch cook to an almost clear suspension, the starch is not in solution. This can be seen by examining the starch cook with a phase contrast or interference contrast microscope, where the swollen granules or granule fragments are evident. Most starches start to swell at about 140-1700F (60-700C), and appear to be reasonably dispersed at 2030F (950C). In order to completely solubilize the starch a much higher temperature of 300-3200F (150-1600C) is required. Some granules in the high amylose starches tend to retain their polarization crosses above 2120F (1000C), but at the 300-3200F range these granules are also completely dissolved.

The cook texture of many starches can be explained by their composition. Waxy corn starch, which is all amylopectin, when heated at 8% starch solids in water becomes very heavy and cohesive as the granules swell. With continued heating, the viscosity decreases as the granules are broken up. On cooling, the viscosity increases. The texture remains cohesive, and the solution retains its clarity. This behavior is typical when the amylopectin fraction dominates the cook characteristics. Regular corn starch, at the same solids, is much thinner, behaves like a short paste, and is quite cloudy when hot. On cooling, the cook becomes on opaque, rigid gel. In this case the amylose has modified the hot cook characteristics of the cooled cooked suspension. If the amylose has too high a molecular weight, then the gelling on cooling may not take place. Tapioca, potato, and sago starches behave more like waxy corn than corn, even though these are amylose containing starches. These amylose containing starches will set to firm gels if the starch is degraded with acid (see fluidity starches). Many of the characteristics of common starches after cooking are given in Table 2.

Table 2. Cook Characteristics of Native Starches (cooked 1 part in 15 parts water at neutral pH).

<table>
<thead>
<tr>
<th>Starch</th>
<th>Hot Cook Body</th>
<th>Hot Cook Viscosity</th>
<th>Viscosity on Prolonged Cooking</th>
<th>Gel Formation on Cooling</th>
<th>Clarity (cold)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Corn</td>
<td>Short</td>
<td>Medium</td>
<td>Stable</td>
<td>Very high</td>
<td>Opaque</td>
</tr>
<tr>
<td>Wheat</td>
<td>Short</td>
<td>Relatively low</td>
<td>Stable</td>
<td>Very high</td>
<td>Opaque</td>
</tr>
<tr>
<td>Amioca</td>
<td>Stringy-cohesive</td>
<td>Moderately high</td>
<td>Thinning</td>
<td>Note</td>
<td>Fairly clear</td>
</tr>
</tbody>
</table>
Likewise, many of the adhesive properties can be explained by reference to the amylose and amylopectin properties of the starch used. Jelly gums are usually made from waxy starches (100% amylopectin) and are stable at room temperature for many months, as would be expected from the slow retrogradation rate of this starch. A corrugating formulation must set to form a bond and become water resistant in a short period of time. The low molecular weight portion of the amylose would be expected to come out of solution fairly rapidly to make a temporary bond, and the high molecular weight portion would more slowly tend to create water resistance.

**MODIFICATION OF STARCHES**

The previous discussion was concerned almost exclusively with the use of native starches dispersed in water. The properties of the dispersions can be changed considerably by additives, or by modifying the starch. The principal modification in the adhesive industry is to decrease the molecular weight of the starch components so as to permit a higher solids content in the formulation. The starch is usually treated in granular form.

**Fluidity Starches**

Fluidity starches are made by hydrolyzing the starch in dilute acid below the gelatinization temperature of the starch. The range of fluidities is 20-90, with a 90 fluidity starch being very much thinner than a 20 fluidity. Fig. 2 gives an indication of the concentrations required to produce a given hot viscosity.

**Oxidized Starches**

A second modification to reduce viscosity is to treat the starch with chlorine under alkaline conditions. Lightly chlorinated starches use the water fluidity method of viscosity designation, but more heavily chlorinated products use a borax fluidity procedure. Chlorinated starches are generally called oxidized starches. They are anionic, in that they contain negative charges. This can be verified by staining with cationic dyes. Methylene Blue is a typical cationic dye.

**Dextrinization**

A third method of reducing viscosity is by the use of dry heat, usually in the presence of acids. This process produces dextrins. Their viscosity is also measured by using the BF or borax fluidity method. Dextrin viscosities are illustrated in Figs. 2, 3, 4. Other dextrin properties are given in Table 3.

**Hydroxyethylation**

Starch is treated with ethylene oxide under alkaline conditions. The major purpose is to slow down retrogradation, or precipitation of amylose from solutions of starch. This is effective because the amylose tends to be uniformly substituted, whereas amylopectin is substituted mainly near branch points.

**Cationic Starches**

Starch is reacted with tertiary or quaternary amine halides or epoxides under alkaline conditions. The cationic starches improve sheet strength, possibly through ionic bonding with slightly anionic paper. The cationic starches are also used in the bottle-labeling adhesives and in paperboard.
<table>
<thead>
<tr>
<th>Dextrin</th>
<th>Acidity</th>
<th>Moisture</th>
<th>Temperature</th>
</tr>
</thead>
<tbody>
<tr>
<td>(DPn)</td>
<td>Color</td>
<td>Swelling</td>
<td>Stability</td>
</tr>
<tr>
<td>White</td>
<td>High</td>
<td>High</td>
<td>Low</td>
</tr>
<tr>
<td>20</td>
<td>White</td>
<td>Partial</td>
<td>Limited</td>
</tr>
<tr>
<td>Yellow or canary</td>
<td>Low</td>
<td>Low</td>
<td>Moderately high</td>
</tr>
<tr>
<td>20-50</td>
<td>Light</td>
<td>High</td>
<td>Good</td>
</tr>
<tr>
<td>British gums</td>
<td>Very low-none</td>
<td>Very low</td>
<td>Moderately high to high</td>
</tr>
<tr>
<td>Very wide range</td>
<td>dark brown</td>
<td>Yellow to complete</td>
<td>Partial to Good</td>
</tr>
</tbody>
</table>

**Amphoteric Starches**

Cationic starches are phosphorylated by heating with orthophosphate to produce a starch with amphoteric (both cationic and anionic) properties. These starches offer improved pigment retention and dry strength over a wide pH range.

**Miscellaneous Derivatives**

These include cross linked or inhibited starches, hydroxypropylated starches, phosphorylated starches, starch succinates, grafted starches, and carboxymethyl starch. Most of the formulations using starch adhesives appear to be obtained by trial and error. The preceding listing of modifications and references is intended to enlarge the scope of possibilities for adhesive development. The discussion of the effects of amylose and amylopectin on starch dispersions serves a similar purpose.

**EFFECT OF ADDITIVES**

**Sodium Hydroxide**

Sodium hydroxide (also called caustic soda) will increase tack, tend to solubilize the starch to a greater degree, increase viscosity, increase cohesiveness, and also increase color. It is usually added after cooking in water, usually to the extent of about 0.5% based on total solids. The increase in pH tends to place negative charges on the starch, which tends to explain dispersion and higher viscosity.

**Borax**

Borax (sodium tetraborate decahydrate) and sodium metaborate (essentially a mixture of borax and sodium hydroxide) change the properties of cooked starch dramatically. There is a large viscosity increase with increasing borax addition to about 15% (based on starch present); tack and cohesiveness are also increased greatly. Borax is used up to 10% based on starch and is usually added before cooking the starch. It acts by complexing with the starch to produce negative charges, and in addition tends to crosslink the starch. It is the crosslinking that causes the large increases in solution viscosity.

**Urea**

Urea is a plasticizer that acts by forming solid solutions with starch and dextrins. The urea tends to prevent crystalization of the starch or dextrin when a film is dried. It is used at 1-10% based on starch present. Other chemicals with similar effects are sodium nitrate, dicyanamide, salicylic acid, thiocyanates, iodides,
gunaidinium salts, and formaldehyde. (Formaldehyde can also crosslink under acid conditions or be used as a preservative).

Glycerol
Glycerol acts as a plasticizer by slowing drying time, preventing excessive drying of a film. In this capacity it is a humectant. Other humectants are ethylene glycol, invert sugars, d-glucose, and sorbitol.

Soluble Soaps
These are used as lubricants to impart flexibility regardless of atmospheric conditions. Too much lubricant will weaken adhesive bonds. Other lubricants are sulfonated castor oil and sulfated alcohols.

Urea-Formaldehyde Resin
This resin is added to give water resistance. Resorcinol-formaldehyde, poly (vinyl acetate), acrylcs, and poly (vinyl alcohol) are also used to increase water resistance.

Miscellaneous Additives
Clays and bentonites are used as fillers in adhesives. Sodium bisulfite, hydrogen peroxide, sodium perborate serve as bleaches. Solvents are added to help wet water-repellent surfaces. Preservatives prevent foam during cooking. Colloid stabilizers such as soaps and sodium chloride are often added.

STARCH ADHESIVES
To be useful as an adhesive, starch must be dispersed in water, usually hot water, and a number of chemicals are added to modify the properties of the starch dispersion. The dispersion and formulation can be done by the user from the different types of starches of modifications of starches previously described. Or the user may purchase formulated adhesives from adhesives manufacturers.

Jelly Gums
These are used for bottle labeling. They are prepared by treating waxy starch or waxy fluidity starch with caustic under high shear. The starch swells in the caustic, and most of the caustic is neutralized with nitric acid. One formulation yields a final composition of 39% starch (40 fluidity waxy), 3% urea (an additive), 3% sodium nitrate (from the sodium hydroxide and nitric acid used), and 56% water (Ref. 10, p. 605). The viscosity as used is about 100,000 cP. The labels are cold water resistant.

Other Liquid Formulations
Alkaline starch formulations (prepared by adding enough caustic to gelatinize the starch) are used as carriers in corrugating, in foil-to-paper adhesives, and in carton or case sealing adhesives. If the caustic used to gelatinize the starch is neutralized, the adhesives at 18-25% starch solids are usable for wall paper printing or foil-to-paper lamination. Combinations of starches and water with added salts are the simplest adhesives, and can be used in bill posting, bag making, and tobacco seam gluing.

Pastes
Short, soft, nonstringy adhesives formulated from heavy bodied starches or dextrins are called pastes. A typical formulation for library paste is to cook a mixture of 45% low-soluble white dextrin, 5% corn starch, 5% glycerol, and 45% water.

Borated Dextrins
Dextrins are frequently formulated with borax, sodium metaborate, boric acid, or caustic in different proportions to give good tack and higher stable viscosities at moderate concentrations. The pH of the formulations is about 9.0 in most cases. Borated dextrins find use in case sealing, carton sealing, tube winding, and laminating.

White Dextrins
White dextrins are prepared by heating dry starch containing relatively large amounts of acid at low temperatures (2580F, 1200C) for rather short times (3-7 hours). Their color is white to cream, and their solubility and viscosity varies from low to high. They are used in bag-seam, tube winding, case and carton
sealing, laminating, gummed sheets, label, and envelope back seam adhesives.

Canary Dextrins
Canary dextrins are prepared by heating dry starch containing a moderate amount of acid at moderate temperatures (3000°F, 1490°C) for a moderate time (11 hours). They are very light to dark tan in color, have a high cold water solubility, a low stable viscosity, and excellent remoistening ability. They are used in gummed tape, envelope front seals, stamps, case and carton sealing, laminating and tube winding.

British Gums
British gums are prepared by heating dry starch with a low amount of acid at high temperature (3300°F, 1660°C) for a long time (17 hours). They have low to high solubilities and low to high stable viscosities, dark color, and fast tack. They are used in solid fiber laminating, bag-seam pastes, and winding adhesives.

Waxy Starch Dextrins
Waxy starch dextrins are similar to dextrins made from non-waxy starch, but have a greater viscosity stability than the corresponding regular dextrin. Envelope front seals, stamps, and gummed sheets are some uses.

Dextrin/Silicate Blends
Fast tack, rigid films, low viscosity, and very good adhesion are obtained from this combination. High-speed case sealing and fiber foil cans are some uses.

Pregelatinized Starches
Pregelatinized starches are prepared by heat swelling starch-water slurry, drying, and grinding the dry powder. The starch at 40% solids is fed to a steam-heated drum where it is gelatinized and dried, and finally scraped off the drum with a blade. These starches are used in some one-tank corrugating applications, multiwall paper bags, wallpaper, and billposting.

APPLICATION AREAS

Papermaking
Cationic, anionic, and amphoteric starch derivatives are cooked and added at the wet end of the papermaking machine at 5-20 pounds per ton of paper. The starch flocculates pigments and fine pulp particles, improves retention of fines, helps drainage, and increases internal (Scott bond) and burst (Mullen) strengths. The starches are retained by the pulp because the pulp has a slight negative charge, which holds the positively charged starch. The positive charge is present directly in the cationic and amphoteric starches, and is formed by complexing with alum in the case of anionic starches. Unmodified starch is also used at 20-40 pounds per ton of pulp. It is retained mainly by entrapment. Although starch is largely added at the wet end of the paper machine, it can be placed directly on the formed sheet. The starch can be sprayed onto the fibers, applied as foam directly on the wet web. These alternative methods have certain advantages, but they lose the flocculating ability of cationic or amphoteric starches.

Paper Coating
In addition to adding starch at the wet end to hold the fibers together, starch can be added at the size press, at the size press with a pigment, at the calender stack, or as a pigmented coating as a separate operation. In all of the above applications, starch acts to bind the paper fibers together or to bind the pigment particles together or to bind the pigment to the fiber. The paper has been partially dried at the size press. The viscosity of the starch solution must be relatively low (50 cP) at the size press. The 2-12% solids used in coatings would have too high a viscosity. Therefore a viscosity reduction is usually necessary. This can be accomplished by enzymes or heat treatment in the case of regular starches, or oxidized, hydroxyethylated, acid fluidity, or acetylated acid fluidity starches can be used. Oxidized starches are anionic, and if recycled may cause pigment and fines retention problems in the wet
Cationic starch which has been preconverted to a low viscosity appears to have advantages in physical properties of the paper and in retention of the surface sized broke. Starch is added at the calender stack for curl control, surface strength, laying of surface fuzz, clay coating holdout, printing characteristics, and grease or oil resistance. Solids vary from 2 to 24%, depending on the starch type used and on the paper requirements. Low viscosity starches, hydroxyethylated converted starches, and oxidized starches are used in this application.

Similar starches to those used on the size press can be used in pigment binding applications, but the viscosity of the starch should be lower. A possible formulation for a 59% solids coating is 0.2% sodium hexametaphosphate, 10.1% calcium carbonate, 40.5% clay, 8.1% low viscosity starch, 0.05% pine oil, and 0.21% soap. Thin boiling cationic starch has also been recommended for this application. The same starches as are used above can be used when starch is added with pigment at the size press. Solids are generally 30-40%, and the starch to pigment ratio can be as high as 1 to 1. The viscosity is usually below 300 cP.

**Corrugating**

Most of the corrugated board is made using the Stein Hall system, which consists of a mixture of gelatinized starch (called the carrier starch) and ungelatinized starch. Flat corrugating medium is fluted, starch is applied to the fluted tips, and a heated liner is brought in contact with the fluted tips under heat and pressure to produce a single facer. Adhesive is applied to the fluted tips on the other side of the corrugating medium and a flat liner called a double backer liner is applied. When the facers are brought in contact with the flutes containing the adhesive, the ungelatinized starch gelatinizes. This creates an extremely high viscosity at the flute-liner interface. The applied heat also evaporates water and further increases the flute-liner bond, so that the corrugated board can be cut without delaminating. The formulation includes sodium hydroxide and borax to decrease the gelatinization temperature of the raw starch and to increase viscosity at the flute after gelatinization.

A typical old formulation is: put 13 parts water, 3.2 parts starch, 0.54 parts caustic dissolved in 0.8 parts water into a tank (called tank 1, or the upper tank), heat with steam to 1600F, agitate 15 minutes, and then add 16 parts cold water. This gelatinizes the starch present and makes the carrier starch portion of the adhesive. In another tank (called tank 2, or the lower tank) 49 parts cold water is mixed with 0.54 parts borax and 18 parts starch is added. The contents of tank 1 are slowly added to tank 2 with efficient mixing. It is possible to carry the entire operation out in tank 2, but control is more difficult.

So called no-carrier or single-component systems depend on careful control to gelatinize the correct proportion of the granules present. One formulation, which increases water resistance by adding urea-formaldehyde, is: mix 20 parts starch, 77 parts water, 0.8 parts 50% caustic, and stir at 1010F until the viscosity reaches 25 Stein Hall seconds. Then add 0.04 parts alum, 0.4 parts boric acid, and 3 parts 60% urea-formaldehyde resin. Other ways to make "no-carrier" systems are to add exactly the correct amount of caustic and stop the swelling by carefully injecting steam into the mixture; or partially swell the granules by milling a 10-40% moisture starch to damage granules before suspending in water. A chemical-mechanical method of producing the carrier starch is to introduce a mixture of 12% starch and 30% caustic into a centrifugal pump. Final alkalinity was 14.4% and the viscosity was 4200 cP. at 760F. The use of starches having different gelatinization temperatures is given in Ref. 14. Here 350 parts tapioca starch and 3000 parts corn starch are suspended in 9000 parts water and treated with 500 parts of borax are added. The Stein Hall viscosity is 52 seconds, and the gel point is 630C (1450F). The tapioca starch swells before the corn starch, and essentially produces the two-component system.

High amylose starches improve water resistance and increase speed of corrugation. A formula for the carrier starch is: 1192 parts water, 424 parts high amylose starch, 6 parts borax: bring to 1300F (540C) and add with stirring a mixture of 36.6 parts caustic in 47.5 parts water. The raw starch portion is made by
mixing 3480 parts water, 1600 parts corn starch, 28 parts borax and 91.2 parts thermosetting resin. Mix the carrier into the raw starch. It appears, at least in the case of high amylose carrier starch, that the carrier starch is the principal adhesive. The raw starch, when gelatinized, absorbs water to concentrate the dispersed high amylose starch on the flutes. Older ideas of the mechanism were that the carrier starch was simply a suspending agent for the raw starch, which was the principal adhesive. One suggestion was that amylose migrating out of the granules caused at least the initial tack.

There are many publications on the use of high amylose starches. High amylose starch is esterified with acetic anhydride or succinic anhydride to improve stability. Another patent concerns the addition of dihydroxyethyleneurea together with acetone-formaldehyde to improve water resistance. The use of high amylose starches in the production of cold corrugating adhesives (not requiring extensive heating) has been published. A mixture of oxidized and hydrolyzed 70% amylose containing starches is dispersed as a 35% starch suspension, gelatinized at 140°C, and used in corrugating without steam. The high amylose starch was treated with 4% sodium hypochlorite at an initial pH 11 to make part A of the degraded mixture. Part B was made by hydrolyzing high amylose starch for 12 hours at 50°C (122°F) with 6% of 35% hydrochloric acid. Both A and B were neutralized to pH 5, washed, filtered, and dried. Then 70% A and 30% B were mixed to a 35% suspension and cooked at 140°C. The use of the B portion is claimed to improve speed from 90 m/min to 230 m/min.

Mixtures of waxy, regular, and high amylose starches are degraded with a mixture of sodium persulfate, sodium sulfate, boric acid, and caustic, and cooked at 90°C (194°F) and about 33% solids to form a starch adhesive that hardens on cooling. The cold corrugating process is described in other references. Other claims for improving corrugating formulations include increasing the carrier starch solids, addition of urea, using crosslinked starch in the carrier, and using cationic starch as the carrier starch.

Bag Adhesives

Three adhesives are used in paper bag manufacture: side seam adhesives, bottom paste adhesives, and cross pastes. The side seam adhesive is used to form a cylinder from a flat sheet of paper. This adhesive must develop a strong bond quickly, so the tube can be cut for further operation. The viscosity should be about 3,000 cP. and solids about 25%. One formulation is: water 68%; heat to 160°F (71°C), add 3% preservative.4 A water-resistant formulation is: water 1700 pounds, white dextrin 700, soap 2, urea-formaldehyde 70, heat at 200°F (930°C), dilute with cold water to 260 gal volume, and add 14 pounds ammonium chloride. The formulation should be used immediately. Its pH is about 6.

Bottom paste adhesives are applied to one end of the tube formed above to close that end to form a bag bottom. These pastes are usually made from unconverted starches. Soap and/or salt may be added to produce a thixotropic paste (flows under shear but sets up if left undisturbed). A water-resistant formulation is corn starch 13%, poly (vinylalcohol) 4.5%, poly (vinylacetate) 1%, soap 0.1%, water 81%, heat to 900°C (1940°F), cool to room temperature.10

Cross paste is used for multiwall bags to glue the plies together before forming a tube. This is similar to seam paste, but should not penetrate the ply. Clay or poly (vinylacetate) is added to prevent penetration.10,11 The seam adhesive for multiwall bags can be somewhat heavier than given above, and higher molecular weight white dextrins can be employed. The bottom adhesive is usually a fluidity starch. A mechanical way of converting potato starch for use in paper bags is also described.29

Laminating Adhesives

The requirements of the particular equipment must be met for bonding paper to paper or to paperboard, for making poster displays, bonding paperboard to paperboard, or rotary lamination. Lay-flat, or noncurling, is an important attribute of these adhesives. High tack and low penetration are also required. One formulation is water 43%, high soluble white dextrin 21%, corn starch 4%, sodium nitrate 32%; heat to 2000°F (930°C),
hold 20 minutes, add preservative. Another formulation is high soluble white dextrin 20%, clay 13.5%, urea 6.7%, borax 5%, water 55%.10

Foil laminations usually call for resins, but even here a small amount of starch is often added for its smoothing properties. One example is: poly (vinylalcohol) 3, starch 3, water 49, potassium persulfate 0.1; add dropwise a mixture of 5 dibutyl phthalate and 39 vinyl acetate at 700°C (1580°F), dilute to 25% solids. Starch at 3-15% is claimed to prevent coarse particle formation in the following formulation: 45% aqueous emulsion containing 1:99 acrylic acid: vinyl acetate copolymer 100, corn starch 4, poly (vinylalcohol) 5, dibutyl phthalate 1.5 Aluminum foil was coated on paper at 300 m/min (meters per minute) without coarse particle formation, while a similar formulation omitting the starch formed coarse particles at 90 m/min.

**Tube Winding**

Tube winding is either spiral (a continuous winding where adhesive is applied to outer plies as they are wrapped on a cylindrical mandrel), or convolute (where the sheet is as wide as the mandrel is long, and the mandrel wraps the sheet over itself). The tube is removed from the mandrel by a pusher arm. Convolute adhesives are usually used cold, while spiral adhesives may be used at 1310°F (550°C). Many starch products may be used, but 50% solids borated dextrins are common.

**Corrugated Boxes**

The tops and bottoms of corrugated boxes are closed with case sealing adhesive. Although liquid glues and hot melts are preferred, carton adhesives are used, sometimes with added caustic. Carton sealing involves bonding the bottom and top flaps of folded paper boxes. A top bottom carton sealing adhesive is water 51%, white dextrin 37%, preservative 1%, borax 6%, antifoam 0.06%; cook to 1850°F (850°C) for 20 minutes, cool to 1200°F (490°C), add 5 water and 0.6 of 50% caustic.

**Gummed Tapes**

The types of remoistenable tapes are regular sealing tapes, reinforced sealing tapes, and box tapes. A regular sealing tape formulation is: thin boiling waxy sealing tape formulation is: thin boiling waxy starch 39.5, canary dextrin 17, polyacrylamide 2, dispersing agent 0.4, water 41.1 is suggested.10 Box tape is usually made from animal glue, but one patent claims performance superior to animal glue from a starch acrylamide graft copolymer prepared as follows: water 51%, sodium nitrate 7%, waxy corn starch 33%, copper sulfate 0.03%, acrylamide 10%; mix, then add 0.05% ammonium persulfate and 0.03% sodium metabisulfate; heat rapidly to 200°F (930°C) and maintain for 15 minutes. Add sodium tartrate to adjust pH to 5.5.

The use of an oxidized starch acetate 180, urea 20, and water 200 as a gummed tape adhesive has been patented. The formulation of an acrylamide starch graft polymer is also claimed: hydrolyzed low viscosity acrylamide-starch graft copolymer 67, canary dextrin 20, animal glue 10, urea 10, petrolatum 0.25, and sodium hexametaphosphate 0.1, are heated for 3 minutes at 185-2050°F (85-960°C).

**Label and Envelope Adhesives**

The general adhesives for these uses can be found in Table 4. Dextrins, enzyme-converted starches, or mechanically degraded starches are suitable, with white or canary dextrins made from tapioca, waxy corn, and regular corn starches the most used. The envelope front seam is remoistenable and can have 55-65% solids with a viscosity of 2,000-10,000 cP. After drying the adhesive should be non-blocking at high humidity. This requires a nonhygroscopic plasticizer. One formulation containing poly (ethylene glycol) is high soluble waxy corn dextrin 63, sodium bisulfite 1, Carbowax 4,000 (Union Carbide) 0.5, water 35.5.

Backseam adhesives can be a little thinner with viscosities close to 1000 cP, and 40-50% solids. Discrete particle, flat gumming, solvent-based adhesives are another method of making remoistenable adhesives. The cold-water-soluble material (usually a dextrin) is suspended in a solvent and applied to the paper and dried. Since the dextrin is not soluble in the solvent, it is deposited in discrete particles. On contact with water, the dextrin swells and will glue two pieces of paper together. The dextrin is usually used
Paper Box

Paper box manufacture involves the operations of ending (box body formation), stripping (coating the box walls with precut wrap and smoothing), tight wrap (machine wrapping a box with glue on most of the wrap), and loose wrapping (adhesive only on border of the wrap). Borated dextrins containing 10-15% borax and cooked at 30-50% solids are usually used. See Table 4 for more information.

Textiles

Starch is used predominantly in warp sizing in the textile field. The purpose is to strengthen the yarns for weaving operations. A typical formulation is corn starch 8.5%, softener (tallow or sulfonated oil) 1.5%, kerosene 0.2%, cook to 1900F (880C), and homogenize at 2000 psi. Oxidized starches are used in finishing and printing, and low treatment starch acetates, starch monophosphate blended with poly (vinyl alcohol), cationic and amphoteric starches, as well as fluidity starches are used in finishing and warp sizing.

Wall Covering Adhesives

These require good wet tack and good slip. Plasticizers give slip (the ability to move the paper around to obtain a good fit), borax gives tack, and clay allows easier removal of the paper later. A formulation consists of acid modified pregelled starch 25, clay 20, urea 3.75, sodium metaborate 1.25, and water 50. A repolymers is: dry blend carboxymethyl cellulose 25, hydroxyethyl cellulose 8.5, sodium alginate 0.5, anionic polyacrylamide 0.3, urea 30, potato starch 27, colloidal silica 1, sodium stearate 2.5, sodium soap 3.5, sodium fluoride 0.5, and 4-chloro-meta-cresol 1. The mixture is suspended in 3 parts water, coated on paper and dried. A second formulation is: starch 375, carboxymethyl cellulose 400, urea 200, polyacrylamide 5, colloidal silica 5, parachloro-meta-cresol 10, sodium fluoride 5. An inverted emulsion formulation containing a starch graft copolymer has also been suggested for wallpaper adhesives.

Miscellaneous Uses

Gelatinized starch 100, dextrin 100 is used to make fish good stable in water. A 1:1 mixture of starch and polyacrylamide was used to make and adhesive for high voltage transformers. A mixture of poly (vinyl acetate) emulsion 100, plasticizer 20, and starch 20 has been used in book binding.

GOVERNMENTAL REGULATIONS: ADDITIVES

The food and Drug Administration (FDA) of the U.S. Department of Health, Education and Welfare (HEW) has issued a series of regulations concerning the use of potentially toxic substances in packaging materials contacting food. Most of the regulations may be found in the Code of Federal Regulations, Title 21, revised annually (21 CFR 121.2520, "Adhesive").